# Dynamics of Light-Induced Conformational Changes of the Phoborhodopsin/ Transducer Complex Formed in the n-Dodecyl $\beta$ -D-Maltoside Micelle

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ABSTRACT: A complex of photoreceptor phoborhodopsin (ppR; also called sensory rhodopsin II) and its cognate halobacterial transducer II (pHtrII) existing in the plasma membrane mediates the light signal to the cytoplasm in the earliest step of negative phototaxis in Natronomonas pharaonis. We have investigated the dynamics of the light-induced conformational changes of the ppR/pHtrII(1-159) complex formed in the presence of 0.1% n-dodecyl  $\beta$ -D-maltoside (DDM) by a fluorescence resonance energy transfer (FRET) based method. Fluorescence donor and acceptor dyes were linked to cysteine residues genetically introduced at given positions in pHtrII and ppR. The light-induced FRET efficiency changes for various pairs of dye-labeled cysteine residues were determined to examine dynamics of movements of given residues in the transmembrane and the linker region including the HAMP domain in pHtrII induced by photoexcitation of ppR. Upon flash excitation of ppR, FRET efficiency changed depending on pairs of the labeled cysteine residues. The distances between V185 in ppR and the five given residues (102 through 141) in the pHtrII linker region estimated from the FRET efficiency increased by 0.3–0.8 Å; on the other hand, the distances between S31 in ppR and the five residues in pHtrII decreased. The changes arose within 70 ms (the dead time of instrument) and decayed at a rate of  $1.1 \pm 0.2$  s. Azide significantly increased the decay rate of light-induced FRET efficiency changes by accelerating the decay of the M state of ppR. The decay rate of FRET efficiency changes coincided with the rate of recovery of the ppR to the initial state but not the decay of the M state. We conclude that the light-induced conformational change of pHtrII occurs before, at the formation or during the M state, and its relaxation is coupled tightly with the decay of the O state of ppR in the 1:1 complex formed in the DDM micelle.

Phoborhodopsin (ppR;<sup>1</sup> also called pharaonis sensory rhodopsin II, NpSRII) is a photoreceptor protein of negative phototaxis in Natronomonas pharaonis. The ppR and its cognate transducer (Halobacterium transducer II; pHtrII) form a 2:2 complex in which the pHtrII dimer is sandwiched between two ppR molecules in the plasma membrane (1, 2). The pHtrII is a homologue of the chemotaxis receptor (methyl-accepting chemotaxis protein: MCP) of Escherichia coli and is thought to be accommodated with a histidine kinase CheA and adaptor protein CheW in the cytoplasmic region. The ppR/pHtrII complex mediates the light signal to the cytoplasmic signal transduction pathway in the earliest step of phototaxis. The structure of the transmembrane region of the complex was determined by X-ray crystallography with atomic resolution (2, 3). The cytoplasmic region of pHtrII has been thought to form a four-helical-bundle coiledcoil structure similar to chemoreceptors (4, 5). The structure of the pHtrII linker region, which connects the transmembrane and cytoplasmic regions, has been investigated by NMR (6, 7), EPR (8, 9), and tryptophan FRET (10) but not determined crystallographically. The *p*HtrII linker region contains the HAMP domain, of which structure and function in the signal transduction process were recently suggested in *Archaeoglobus fulgidus* by NMR (11).

The dynamics of ppR photointermediates (states) has been well investigated. Photoexcitation of ppR induces a cyclic conformational change (photocycle), in which K, L, M, and O intermediate states named in the order of their appearance were identified (12-14). On the other hand, the conformational change of pHtrII in the ppR/pHtrII complex induced by illumination of ppR has less information; Klare et al. showed a rotation of transmembrane helix 2 (TM2) of pHtrII, which was induced by the outward tilt of helix F of ppR, during the M  $(M_2)$  state of ppR (1, 15), and Yang et al. showed that the membrane proximal region of pHtrII became close to the E-F loop of ppR when it was photoexcited (10). These conformational changes are thought to occur without secondary structure change, suggesting rigid-body motion of the complex (16). However, conformational changes of the linker region and the cytoplasmic region of pHtrII during state change in ppR have never been detected directly, although a rotation of the HAMP domain in the linker region was suggested (11). Since the M state has the longest lifetime in the photocycle of ppR and large conformational change is induced during the M state in the case of bacteriorhodopsin

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<sup>&</sup>lt;sup>1</sup> Abbreviations: ppR, pharaonis phoborhodopsin; pHtrII, pharaonis halobacterial transducer II; FRET, fluorescence resonance energy transfer; DDM, n-dodecyl β-D-maltoside; TM, transmembrane; EPR, electron paramagnetic resonance.

(reviewed in ref 17), it has been thought that the M state is the most crucial state in the signaling process of ppR (signaling state). Recently, Sudo and Spudich showed that bacteriorhodopsin modified with only three amino acid mutations in its membrane-embedded domain mediates phototaxis signals through the pHtrII genetically expressed in the Halobacterium salinarum strain which lacks the four native rhodopsins (bacteriorhodopsin, halorhodopsin, sensory rhodopsin I, and phoborhodopsin) and the two transducer proteins (HtrI and HtrII) (18), implying that the conformational change that occurs in the pumping cycle of bacteriorhodopsin shares signal-producing features with that of ppR. In the case of ppR, Spudich and colleagues concluded from experiments using retinal analogues that both M and O states are signaling states (19). On the other hand, studies using a laser flash photolysis and transient grating method indicated that the dynamics of volume expansion and contraction of the ppR/pHtrII complex coincided with the rise and decay of the O state, respectively, suggesting that the O state is the signaling state (20). EPR study indicated that the relaxation of rotation of TM2 in pHtrII was slower than the whole process of the photocycle of ppR, suggesting that conformational change of pHtrII is uncoupled from the state changes of the ppR during the last step of the photocycle (1). Thus, the previous results concerning dynamics of the conformational change of the complex are controversial.

Recently, we employed a fluorescence resonance energy transfer (FRET) based method, which provides a convenient spectroscopic method to study the dynamics of the structural change of the protein complex, to investigate light-induced conformational changes of the ppR/pHtrII complex (21). In the application of FRET to the ppR/pHtrII system, spectral overlap of the dyes with the absorption of the intermediates of ppR per se had been thought to prevent a quantitative analysis. This problem was solved by using recently developed far-red fluorescent dyes, Cy5 and Alexa660 as fluorescent donors and Cy7Q, HiLyte Fluor 750 (HF750), and DY750 as acceptors. Subangstrom-scale changes of distances from V185 in ppR to V78, A79, and D102 in pHtrII were successfully detected using the pair Alexa660 and Cy7Q. In this study, we have investigated the dynamics of the lightinduced movement of several residues in the pHtrII (A80, D102, E110, L120, S130, and A141; see Figure 1) relative to S31, V58, and V185 in ppR and intramolecular movement of A80 against L120 in pHtrII in the ppR/pHtrII complex formed in 0.1% DDM and the effect of azide on the dynamics. The results of FRET experiments suggest that the light-induced conformational change of pHtrII occurs before, at the formation, or during the M state, and its relaxation takes place concomitantly with the decay of the O state of ppR but not the M state.

#### MATERIALS AND METHODS

*Reagents*. The thiol reactive dye Alexa660 C<sub>2</sub>-maleimide was purchased from Invitrogen Corp. (Carisbad, CA). Cy5-maleimide and Cy7Q-maleimide were from GE Healthcare Bio-Sciences Corp. (Piscataway, NJ). The detergents n-dodecyl β-D-maltoside (DDM) and n-octyl β-glucoside (OG) were from Dojindo Laboratories (Kumamoto, Japan). All*trans*-retinal and isopropyl 1-thio-β-D-galactopyranoside (IPTG) were from Wako Pure Chemical Industries, Ltd. (Osaka, Japan).

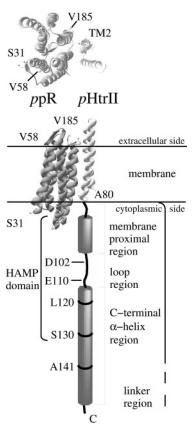


FIGURE 1: Scheme of positions for single-cysteine substitution in ppR and pHtrII examined in this study. The top is a view from the cytoplasmic side. The ribbons represent the crystal structure of the complex (PDB code 1H2S). The cylinders and the lines in the pHtrII linker region were predicted  $\alpha$ -helix and extended (loop) regions, respectively.

Construction of Plasmids. Expression plasmids of ppR and pHtrII(1-159) possessing a  $6 \times$  histidine tag at the C terminus were constructed previously (22), where pHtrII-(1-159) stands for truncated pHtrII protein expressed from the 1st to 159th residue instead of the whole length. Expression plasmids for ppR and pHtrII(1-159) with single cysteine substitution at various positions were prepared by PCR using the QuikChange method. Introduction of the single mutation was confirmed by sequence analysis using an automated sequencer, PRISM 3100-Avant, and a BigDye Terminator v3.1 cycle sequencing kit (Applied Biosystems).

Protein Expression, Purification, and Fluorescent Labeling. Wild-type and mutant proteins of ppR and pHtrII(1-159) were expressed in E. coli BL21(DE3) cultured in  $2\times$ YT medium by IPTG (1 mM) induction (23). Purification of each protein was performed with Ni-NTA agarose (Qiagen) and DEAE-Sephacel (GE Healthcare) columns as described previously (21). The purities of the ppR and pHtrII were checked by SDS-PAGE and, further, the ratio of absorbance at 500 nm to that at 280 nm in the case of ppR. SDS-PAGE of the purified ppR and pHtrII showed a single band corresponding to 25 and 17 kDa, respectively (see Figure 2A). Purified pHtrII and ppR were labeled with donor (Alexa660-maleimide) and acceptor (Cy7Q-maleimide) on their genetically introduced cysteine residue, respectively (21). The concentrations of ppR and dyes were determined by measuring their absorbance, and that of pHtrII was done by using the Micro BCA protein assay kit (Pierce). Labeling efficiency was calculated from their concentrations for each

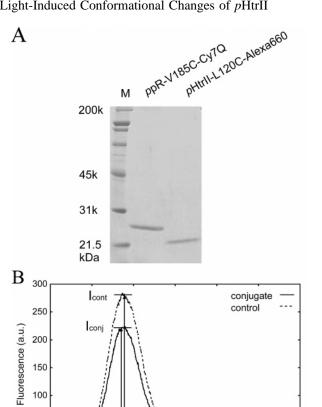


FIGURE 2: (A) SDS-PAGE analysis of dye-labeled protein. Three micrograms of each protein was loaded on 15% polyacrylamide gels. Labeling efficiencies for ppR-V185C-Cy7Q and pHtrII-L120C-Alexa660 were 0.45 and 0.51, respectively. Expected molecular masses of ppR and pHtrII proteins are 25 and 17 kDa, respectively, and those of Cy7Q and Alexa660 are 1.1 and 0.9 kDa, respectively. (B) Fluorescence spectra of the ppR-V185C-Cy7Q/ pHtrII-L120C-Alexa660 complex (conjugate, solid line) and the nonlabeled ppR-V185C/pHtrII-L120C-Alexa660 complex and free Cy7Q at their corresponding concentrations (control, dotted line). The concentration of ppR and pHtrII was 8 and 4  $\mu$ M, respectively. The excitation wavelength was 610 nm.

750

Wavelength (nm)

800

850

900

50

600

650

sample preparation. The labeling efficiencies of acceptor to ppR-S31C, -V58C, and -V185C were 0.37-0.58, 0.86, and 0.68–0.91, respectively, and those of donor to pHtrII-A80C, -D102C, -E110C, -L120C, -S130C, and -A141C were 0.25-0.28, 0.48-0.53, 0.51-0.70, 0.38-0.59, 0.38-0.57, and 0.57, respectively. For double labeling of pHtrII using A80C and L120C by the donor and acceptor dyes, the protein was incubated with 3-fold Alexa660-maleimide and 6-fold Cy7Qmaleimide for 2 h at room temperature. Their labeling efficiencies were 0.30 and 0.71, respectively. This sample should be a mixture of nonlabeled, differently single-labeled, and double-labeled pHtrII. However, among them, only pHtrII with A80<sub>donor</sub>-L120<sub>acceptor</sub> or A80<sub>acceptor</sub>-L120<sub>donor</sub> are expected to exhibit the flash-induced change in FRET efficiency. No degradation of the proteins during the dyelabeling process was confirmed as shown in Figure 2A.

Measurement of FRET Efficiency. The procedure for measurement of FRET efficiency and analysis of the data were done as described previously (21). Prior to the fluorescence measurements, samples were incubated in buffer M (50 mM HEPES, pH 7.0, 400 mM NaCl, 0.1% DDM)

for at least 3 h to form the complex. To obtain FRET efficiency, the effect of absorption of the fluorescence emitted from the donor by the acceptor was eliminated as follows. The fluorescence spectrum in a solution of the ppR-acceptor/ pHtrII-donor complex (conjugate) was measured concomitantly with that in a solution containing the nonlabeled ppR/ pHtrII-donor complex and free acceptor (control) at their corresponding concentrations, as shown in Figure 2B. The FRET efficiency was determined by analyzing the spectra illustrated in Figure 2B by the equation:

$$E = \frac{1}{\gamma} \left( \frac{I_{\text{cont}} - I_{\text{conj}}}{I_{\text{cont}}} \right) \tag{1}$$

where  $\gamma$  is a labeling efficiency of acceptor and  $I_{\rm conj}$  and  $I_{\rm cont}$  are the fluorescence intensities at the peaks in the spectra of the conjugate and control samples. The FRET-derived distances and their changes were calculated under the assumption that FRET does not occur in the control because the average distance between the donor and the bulk acceptor is sufficiently long compared to the Förster distance.

Time-Resolved FRET Measurement. Transient changes of fluorescent intensity upon photoexcitation of ppR were measured with a fluorescence spectrometer associated with a flash apparatus and analyzed as previously described in ref 21. For the measurements in the presence of azide, the electronic low-pass filter (cutoff frequency: 100 Hz) was removed to reduce the dead time of the instrument. To avoid the presence of labeled pHtrII which is not associated with labeled ppR, 8 µM labeled ppR and 4 µM labeled pHtrII were incubated in buffer M for at least 3 h at room temperature and submitted to the fluorescent measurement. All measurements were carried out at 20 °C, and data were obtained as an average of 160 flash measurements. Apparent FRET efficiency at time t,  $E_{app}(t)$ , was calculated from the fluorescence intensities of the conjugate and control samples at 690 nm (21). The distance change  $[\Delta R(t)]$  was obtained by using the formula:

$$\frac{\Delta R(t)}{R} = -\frac{1}{\psi} \left( \frac{\Delta E_{\text{app}}(t)}{6E(1-E)} \right) \tag{2}$$

assuming  $\Delta R(t) \ll R$ . Here, E is the FRET efficiency at the initial state,  $\Delta E_{app}(t)$  [= $E_{app}(t) - E$ ] is apparent FRET efficiency change at time t, and  $\psi$  is the excitation efficiency of ppR upon the flash, which is dependent on the arrangement of the apparatus. The value of  $\psi$  in each experiment was determined from the relative reduction of the ppR initial state upon the flash obtained by measuring the absorbance change at 500 nm. The values were 0.34-0.52 among all measurements.

Transient ppR Absorbance Change Measurements. The apparatus for measurement of transient absorbance changes during the photocycle was essentially the same as that of the transient fluorescence measurement except for the source of measuring light. An I<sub>2</sub> lamp was used as the source, and a band path filter (Toshiba, KL40, KL50, or KL56) was placed in front of the cell. The intensity of transmitted light was recorded for each wavelength in the absence of the SC64 filter. To avoid the presence of ppR, which is not associated with pHtrII, 4 μM ppR was incubated with 16 μM pHtrII in buffer M for at least 3 h at room temperature (24). All

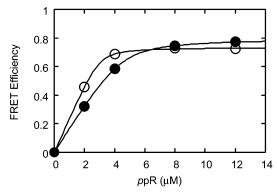


FIGURE 3: FRET efficiencies at various concentrations of ppRV185C under a fixed pHtrII-A79C (open circles) or pHtrII-D102C (closed circles) concentration of 4  $\mu M$ . Cy5 and Cy7Q were utilized as the fluorescence donor and acceptor, respectively.

measurements were performed at 20  $^{\circ}$ C, and the data were obtained as an average of 16 flash measurements. The least-squares curve fitting analysis was performed using Gnuplot software as described previously (12). The time-dependent absorbance changes of the M, O, and initial state of ppR were given as

$$\Delta A_{\rm M} = A_1 \exp(-k_1 t)$$

$$\Delta A_{\rm O} = A_2 \frac{k_1}{k_2 - k_1} [\exp(-k_1 t) - \exp(-k_2 t)]$$

$$\Delta A_{\rm I} = A_3 \frac{1}{k_2 - k_1} [k_1 \exp(-k_2 t) - k_2 \exp(-k_1 t)]$$
 (3)

respectively, where  $k_1$  and  $k_2$  are decay rates of the M and O states and  $A_1$ ,  $A_2$ , and  $A_3$  are constants depending on the measuring wavelength, respectively. The observable absorbance changes at 400, 500, and 560 nm are given by

$$\Delta A_{400} = \Delta A_{M} + \alpha_{1} \Delta A_{O} + \alpha_{2} \Delta A_{I}$$

$$\Delta A_{560} = \alpha_{3} \Delta A_{M} + \Delta A_{O} + \alpha_{4} \Delta A_{I}$$

$$\Delta A_{500} = \alpha_{5} \Delta A_{M} + \alpha_{6} \Delta A_{O} + \Delta A_{I}$$
(4)

where  $\alpha_1 - \alpha_6$  represent the contribution of spectral overlap of these states at their wavelengths. The kinetic data at 400, 500, and 560 nm were analyzed by combining eqs 3 and 4.

# RESULTS AND DISCUSSION

Conformation in the pHtrII Linker Region in the ppR/pHtrII Complex Formed in DDM. The FRET-derived dye to dye distances between each of several residues in the pHtrII linker region (D102, E110, L120, S130, and A141) and each of three residues in ppR (S31, V58, and V185) in the initial state complex were measured. Here, S31 and V58 are located in the cytoplasmic side and extracellular side of the helix B, respectively, and V185 is located in the extracellular F–G loop (see Figure 1). As illustrated in Figure 3 for the pairs ppR-V185C-Cy7Q/pHtrII-A79C-Cy5 and ppR-V185C-Cy7Q/pHtrII-D102C-Cy5, when the concentration of the ppR-acceptor in the solution was increased (0–12  $\mu$ M) at a fixed concentration of the pHtrII-donor (4  $\mu$ M), FRET efficiency increased with an increase of ppR-acceptor concentration and saturated at 4  $\mu$ M, suggesting that ppR

and pHtrII form a complex with a 1:1 ratio in 0.1% DDM. These results are consistent with previous results obtained in detergent solution in which a 1:1 ppR/pHtrII heterodimer complex (1:1 complex) was found to form (1, 9, 24, 25). On the basis of these results, FRET efficiency and its change were measured with the solutions containing 8  $\mu$ M pPR and 4  $\mu$ M pHtrII and analyzed under the assumption of the complete formation of the 1:1 complex. The distances derived from FRET efficiency are summarized in Table 1.

It is evident that FRET-derived distances are different from the corresponding amino acid distance ( $C_{\alpha}$  distance) estimated from the crystal structure because of the bulkiness, orientation, and mobility of the dyes and the microenvironment separating the donor and acceptor dyes, such as the refractive index of the medium. For example, the FRETderived distances from A80 in pHtrII to S31 (located in cytoplasmic side), V58 (extracellular side), and V185 (extracellular side) in ppR were 67, 60, and 56 Å, respectively, which were considerably longer than that of the corresponding amino acid distance in the crystal, 30, 34, and 29 Å, respectively (2). Furthermore, the FRET distance between A80 in pHtrII and S31 in ppR was longer than that between A80 in pHtrII and V58 in ppR, while the amino acid distance between A80 and S31 was shorter than that between A80 and V58 in the crystal (2). This discrepancy may be explained by considering the effect of the refractive index (n) of the medium separating the donor and acceptor dyes: if we assume n = 1.33 (water) for ppR-S31/pHtrII-A80 and n = 1.45 (1.4 for lipid and 1.5 for protein) for ppR-V58/ pHtrII-A80, the Förster distance  $R_0^6$  for the former becames larger by a factor of 1.41 than that for the latter because  $R_0^6$ is inversely proportional to  $n^4$  (26). When the environment of the dyes is substantially equivalent as in the case of the ppR-V58/pHtrII-A80 and ppR-V185/pHtrII-A80 pairs, the difference between the FRET-derived distance for the ppR-V58/pHtrII-A80 pair and that for the ppR-V185/pHtrII-A80 pair (5 Å) was consistent with the corresponding value determined in crystal (4 Å), suggesting that the FRET-derived distance well reflects the amino acid distance. Since S31 in ppR and the residues in the HAMP domain of pHtrII are located in the aqueous environment on the cytoplasmic side, the FRET-derived distances from S31 in ppR to these residues in pHtrII may be compared without this environmental effect. The FRET-derived distances from S31 in ppR to D102 or E110, in the putative loop region of the pHtrII linker, were not so different from that between S31 in ppR and A80 in the transmembrane region of pHtrII, suggesting that the loop region is located in the vicinity of ppR. These results are consistent with the previous EPR-based structural model of the pHtrII membrane proximal region (83–101), in which the region interacts with the E-F loop of ppR, in the detergent solution (9). The FRET-derived distances from S31 or V185 in ppR to L120, S130, or A141 in the second  $\alpha$ -helix region in the pHtrII linker (6–8, 10) increased toward the C terminus in the pHtrII linker, suggesting a rod-like structure of the region, which elongates to a direction going away from ppR. This aspect of the pHtrII linker region in the complex formed in the presence of DDM does not agree with the solution structure of the HAMP domain fragment in the absence of detergent, in which helixes of two HAMP domain fragments were suggested to interact with each other and form the homodimeric, parallel four-helical coiled-coil

Table 1: Quantum Yield (QY) of pHtrII-Alexa660 and Förster Distance (R<sub>0</sub>), FRET Efficiency (E), FRET-Derived Distance (R), and Change of the Distance ( $\Delta R$ ) by Flash Excitation of  $ppR^a$ 

		transmembrane	linker				
	pHtrII	A80	D102	E110	L120	S130	A141
ppR V185 S31 V58	$\begin{array}{c} \operatorname{QY} \\ R_0(\mathring{\mathrm{A}}) \\ E^b \\ R(\mathring{\mathrm{A}})^c \\ \Delta R(\mathring{\mathrm{A}})^d \\ E^b \\ R(\mathring{\mathrm{A}})^c \\ \Delta R(\mathring{\mathrm{A}})^d \\ E^b \\ R(\mathring{\mathrm{A}})^c \\ \Delta R(\mathring{\mathrm{A}})^d \end{array}$	$0.29$ $61$ $0.61 \pm 0.05$ $56 \pm 1.9$ $-1.3 \pm 0.07$ $0.36 \pm 0.002$ $67 \pm 0.1$ $-0.67 \pm 0.08$ $0.52 \pm 0.03$ $60 \pm 1.0$ $1.2 \pm 0.4$	$0.34$ $64$ $0.54 \pm 0.03^{e}$ $63 \pm 1.4^{e}$ $0.33 \pm 0.1^{e}$ $0.36 \pm 0.005$ $71 \pm 0.2$ $-1.8 \pm 0.3$ $0.38 \pm 0.02$ $70 \pm 0.6$ $0.36 \pm 0.03$	$0.29$ $64$ $0.47 \pm 0.05$ $65 \pm 2.3$ $0.54 \pm 0.22$ $0.34 \pm 0.002$ $71 \pm 0.08$ $-1.3 \pm 0.59$	$\begin{array}{c} 0.26 \\ 61 \\ 0.45 \pm 0.11 \\ 63 \pm 4.7 \\ 0.77 \pm 0.06 \\ 0.23 \pm 0.03 \\ 75 \pm 1.7 \\ -0.61 \pm 0.32 \\ 0.10 \pm 0.01 \\ 88 \pm 1.3 \\ \text{ND} \end{array}$	$0.30$ $63$ $0.37 \pm 0.01$ $69 \pm 0.5$ $0.45 \pm 0.04$ $0.20 \pm 0.002$ $80 \pm 0.2$ ND	$0.28$ $63$ $0.31 \pm 0.04$ $73 \pm 2.2$ $0.32 \pm 0.02$ $0.17 \pm 0.01$ $82 \pm 1.0$ ND

<sup>&</sup>lt;sup>a</sup> Data represent the mean  $\pm$  SD of two individual experiments. The residues in ppR were labeled with Cy7Q, and those in pHtrII were labeled with Alexa660. ND = not detected. <sup>b</sup> FRET efficiencies were calculated using eq 1. <sup>c</sup> FRET-derived distances were calculated using  $E = R_0^{6}/(R_0^{6})$  $+R^6$ ) under the assumption of isotropic orientation of dyes, i.e.,  $\kappa^2 = \frac{2}{3}$ , and the refractive index used in calculation of the Förster distance  $R_0$  was 1.4 (26). <sup>d</sup> FRET-derived distance changes upon flash excitation of ppR were calculated using eq 2. <sup>e</sup> Taniguchi et al. (21).

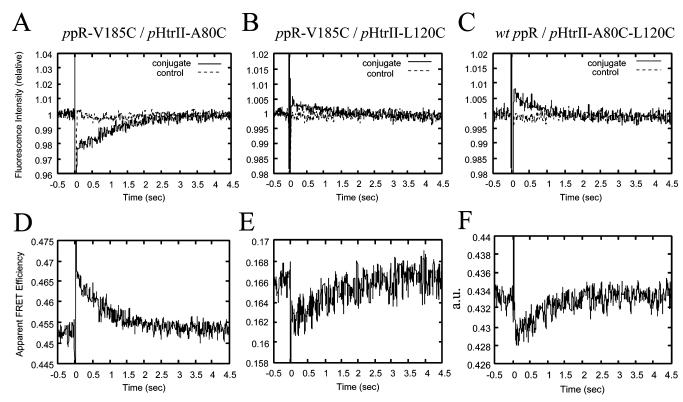


FIGURE 4: Change of FRET efficiency upon a flash excitation of ppR. (A-C) Changes of fluorescence intensity of the conjugate and control samples of ppR-V185C/pHtrII-A80C (A), ppR-V185C/pHtrII-L120C (B), and wt ppR/pHtrII-A80C-L120C (C). The excitation wavelength was 633 nm (He-Ne laser). Fluorescence was monitored at the wavelength giving the maximum of the donor fluorescence (690 nm). Each trace was an average of 160 flash data recorded at 20 °C. The very sharp decrease and recovery within 70 ms was an instrumental artifact. The concentrations of ppR and pHtrII were 8 and 4 µM, respectively. (D-F) Change of apparent FRET efficiency upon a flash excitation of ppR in ppR-V185C/pHtrII-A80C (D), ppR-V185C/pHtrII-L120C (E), and wt ppR/pHtrII-A80C-L120C (F). FRET efficiency was calculated from the data shown in (A) and (B) without correction of labeling efficiency and excitation yield of ppR. In the case of ppR/pHtrII-A80C-L120C (F), the sample was a mixture of nonlabeled, differently single-labeled, and double-labeled pHtrII. However, among them, only pHtrII with A80<sub>donor</sub>-L120<sub>acceptor</sub> or A80<sub>acceptor</sub>-L120<sub>donor</sub> are expected to exhibit the flash-induced change in FRET efficiency.

structure by NMR (11). This discrepancy may be due to the absence of pHtrII-pHtrII dimerization in the 1:1 ppR/pHtrII complex and looseness of the hydrophobic interaction between two helixes in a HAMP domain fragment by the presence of detergent.

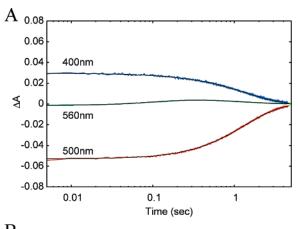
Conformational Change of the Complex upon Flash Excitation of ppR. Flash-induced changes in the FRETderived distances were investigated for various pairs of the ppR and pHtrII residues (Figure 4, Table 1). Upon a flash

excitation of ppR, the FRET efficiency of the ppR-V185C-Cy7Q/pHtrII-A80C-Alaxa660 pair initially increased within 70 ms (dead time of the instrument) and relaxed with a rate of  $1.1 \pm 0.02$  s to the level of the initial state (Figure 4B). In the case of the ppR-V185C/pHtrII-L120C pair, on the other hand, the FRET efficiency initially decreased (Figure 4D). Thus, the difference in direction and magnitude of the FRET efficiency changes induced by a flash excitation of ppR was observed among the pair of residues examined. An intramolecular flash-induced distance change was also observed between A80 and L120 in pHtrII labeled with Alexa660 and Cy7Q. As shown in Figure 4F, the FRET efficiency of ppR/pHtrII-A80-L120-Alexa660-Cy7Q decreased within 70 ms and relaxed at similar rate,  $1.0\pm0.11$  s. Distance changes at various pairs of residues calculated from the FRET efficiency changes after correction of labeling efficiency and flash yield are summarized in Table 1.

The positions of V58 and V185 in ppR scarcely change from the initial (unphotolyzed) at the M state in the crystal at low temperature (3). If we assume that these residues do not move in detergent solution at room temperature, Table 1 indicates that, upon illumination of ppR, the residues in the linker region (D102, E110, L120, S130, and A141) move away from both V58 and V185 in ppR, located at the extracellular surface, by 0.3–0.7 Å. The simplest suggestion is as follows: Conformational changes occur in the  $\alpha$ -helix of the membrane proximal or C-terminal α-helix region where the tight  $\alpha$ -helix becomes slightly loose so as to increase the distance. However, the change of 0.3-0.7 Å seems too small by this suggestion, because the length per one pitch of the  $\alpha$ -helix is 5.4 Å. Another possibility might be a rotation of these regions as pointed out by Klare et al. (1, 15). However, from the present data, we cannot infer whether the rotation occurs or not. The other possibility to be considered is the interaction between the E-F loop in ppR and the membrane proximal region pointed out by Yang et al., suggesting that the membrane proximal region of pHtrII become close to the E-F loop of ppR when it was photoexcited (10). Neither we can infer the validity of this possibility: Lateral (i.e., along with membrane) movement of the membrane proximal region of pHtrII is difficult to detect as a change of distance from V58 or V185 in ppR, and the decrease of the FRET-derived distance from S31 in ppR to the residues in the linker region of pHtrII (D102, E110, and L120) does not necessarily indicate that these residues of pHtrII move to ppR, because S31 in ppR is reported to move toward helix A by 0.7 Å in the crystal (3). Then, the interpretation of the present data is difficult. It should be mentioned, however, that the photoinduced changes in distances from V185 of ppR to A80 and to the residues in the linker region are in opposite directions, suggesting that the conformational changes in the linker region might be different from that of the transmembrane region.

Thus, from the data obtained here, we cannot draw a possible model of the conformational change, but it should be worth noting that the present report reveals the occurrence of the conformational change in the HAMP domain when *ppR*, a receptor, is stimulated.

Coupling of Conformational Changes of pHtrII and ppR. As shown in Figure 4, all of the FRET efficiency changes observed in several pairs of the residues in ppR and pHtrII including an intramolecular residual movement in pHtrII arose within the dead time of the instrument (70 ms) and decayed at a rate of  $1.1 \pm 0.2$  s upon flash-light excitation of ppR, suggesting that it occurred before or at the formation of the M state and relaxed concomitantly with the decay of the M state or that of the O state. To clarify the coupling process between pHtrII and ppR in the relaxation of the light-induced conformation, we investigated the effect of azide on the decay rate of the flash-induced FRET efficiency



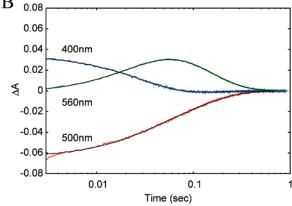


FIGURE 5: Photocycle of ppR in the 1:1 ppR/pHtrII complex formed in 0.1% DDM solution. (A, B) Flash-induced absorbance changes of ppR-V185C/pHtrII-L120C in the absence (A) and presence (B) of 100 mM azide. Absorbance changes measured at 400 nm (mainly M state), 500 nm (initial state), and 560 nm (O state) were shown. The regression curves obtained by least-squares curve fitting analysis (eq 4) were indicated as black lines. The decay rates of M and O states were 0.73 and 8.8 s<sup>-1</sup> in the absence and 34 and 8.4 s<sup>-1</sup> in the presence of azide, respectively. The concentration of ppR-V185C and pHtrII-L120C was 4 and 16  $\mu$ M, respectively.

change and compared them with the rates of the state change of ppR in the absence and presence of azide. Azide has been known to facilitate transfer of the proton to the deprotonated Schiff base in the M state and accelerate the decay of the M state of ppR in detergent solution (27) and in the lipid membrane (28), both in the absence of pHtrII. At first, we investigated the dynamics of the photocycle in the ppR/pHtrII complex formed in 0.1% DDM and the effect of azide on the rates of the photocycle. In the absence of azide, the decay rates of M and O states in the ppR/pHtrII complex were 0.73 and 8.8 s<sup>-1</sup>, respectively (Figure 5A). The addition of 100 mM azide accerelated the M decay by about 50-fold (Figure 5B), which is consistent with the report by Sudo et al. (29). The decay rate of the O state in the presence of 100 mM azide was 8.4 s<sup>-1</sup>, which was almost the same as that in the absence of azide, 8.8 s<sup>-1</sup>. Figure 6A shows the effect of azide on the decay process of the flash-induced FRET efficiency change in the ppR/pHtrII 1:1 complex. The decay rate increased with increasing concentration of azide, suggesting that the conformational change of the pHtrII is tightly coupled with that of ppR in the photocycle. The decay coincided with the recovery of ppR to the initial state but not the decay of the M state in the absence and presence of azide (Figure 6B,C). Essentially the same results were

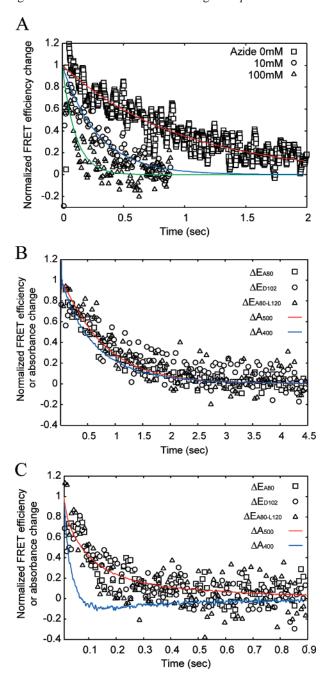


FIGURE 6: Dynamics of the relaxation process of light-induced motion between the indicated residues in the 1:1 ppR/pHtrII complex formed in 0.1% DDM solution. (A) Relaxation of FRET efficiency to the initial level in the presence of the indicated concentrations of azide. FRET efficiency change was measured in the ppR-V185C-Cy7Q/pHtrII-A80C-Alexa660 pair. The solid lines were single exponential curves obtained by a least-squares curve fitting procedure. (B, C) FRET efficiency change overlaid on absorbance changes at 400 nm (blue) and 500 nm (red) in the absence (B) and presence of 100 mM azide (C). FRET efficiency changes were measured in the ppR-V185C-Cy7Q/pHtrII-A80C-Alexa660 (squares), ppR-V185C-Cy7Q/pHtrII-D102C-Alexa660 (circles), and wt ppR/pHtrII-A80C-L120C-Alexa660-Cy7Q (triangles) pairs.

obtained in the transmembrane region (pHtrII-A80C), linker region (pHtrII-D102C), and intramolecular (pHtrII-A80C-L120C) movements. These results indicated that the relaxation of the light-induced conformation of pHtrII to the initial state was tightly coupled with the decay of the O state of ppR.

Sudo et al. estimated the dissociation constant between ppR and pHtrII at the ground, M, and O states to be 0.16  $\mu$ M (30), 15  $\mu$ M (24), and 0.15  $\mu$ M (31), respectively. Combining these values with the present results, we can suppose as follows: before M or at the M formation, the signal transfers from the receptor ppR to pHtrII, and then the interaction between them becomes weak by a factor of 100. At the O state, they rebind strongly, and the change in ppR from O to the original state and the return of the pHtrII conformation to the original dark state are tightly coupled with each other in the ppR/pHtrII complex. It should be mentioned that the decay rate of the O state in the ppR/ pHtrII complex increased by 6-fold compared with that in the absence of pHtrII,  $1.5 \text{ s}^{-1}$  (data not shown), suggesting that the relaxation of the O state of ppR is accelerated by the formation of the complex with pHtrII as to be coupled with the conformational change of pHtrII. The same behaviors were obtained previously in 0.1% DDM solution for the O-like state, which was measured in the M state-lacking mutant, ppR-D75N (31).

Previously, it has been shown with FTIR that the hydrogen bond between Y199 in ppR and N74 in pHtrII re-formed concomitantly with the decay of the M and/or O states in the lipid membrane (32). A study using the transient grating method has shown that the rates of the volume expansion and contraction, which may reflect conformational change, took place concomitantly with the rise and decay of the O state, respectively, in detergent solution (20). However, EPR studies using the 2:2 complex formed in the lipid membrane showed that the light-induced change in an interspin distance between the spin-labeled V78 residues located in the interface of two pHtrIIs occurred at a rate of 3 ms (15), which is in conflict with the data obtained by the transient grating method but not with the present data obtained in the 1:1 ppR/ pHtrII complex formed in detergent solution. Interestingly, however, the relaxation of the light-induced distance change between V78 residues in a pHtrII/pHtrII pair in the 2:2 complex observed by EPR measurements was slower than the recovery of ppR to the initial state by a factor of 4, suggesting decoupling of the pHtrII relaxation from return of ppR to the initial state (1). The present results clearly indicate that the relaxation of the light-induced conformation of pHtrII in the 1:1 ppR/pHtrII complex is coupled tightly with the decay of the O state of ppR both in the absence and in the presence of azide. The discrepancy between the results obtained with the 2:2 complex in the lipid bilayer and the 1:1 complex in detergent solution may depend on the difference in the paired residues examined but not on the experimental technique. In the present FRET experiments, the light-induced conformational change occurring in a single pair of ppR and pHtrII was examined. On the other hand, since EPR experiments examined the relative motion of two pHtrIIs in the 2:2 ppR/pHtrII complex, the observed dynamic behavior reflects relative motion between a pair of 1:1 ppR/ pHtrII units in the 2:2 ppR/pHtrII complex. The relaxation of the light-induced interunit conformation in the 2:2 complex may delay to that of the intraunit conformation.

## **CONCLUSION**

By FRET experiment, the dynamics of light-induced intramolecular motion between A80 and L120 in pHtrII as well as intermolecular motions between each of several residues in ppR and pHtrII in the 1:1 ppR/pHtrII complex formed in DDM micelle were determined. The loop and the second α-helix in the pHtrII linker move away from V185 and V58. However, the structure of the linker region in the 1:1 complex formed in detergent solution may be different from that in the 2:2 complex formed in the lipid membrane. Further studies in the lipid membrane are required. O-decay of ppR became faster in the 1:1 ppR/pHtrII complex formed in DDM compared with that in the absence of pHtrII, indicating that the dynamics of conformational change in ppR from the O state to initial state is modified by the formation of the complex with pHtrII as to be accelerated. Flash-induced FRET efficiency change showed that the relaxation of the light-induced conformation of pHtrII coincides with that of the O state of ppR in the 1:1 ppR/ pHtrII complex, suggesting that return of the light-induced conformation of pHtrII to the initial state is tightly coupled with the conformational change from the O state to the initial state of ppR. Taken together with an EPR study with the 2:2 complex formed in the lipid membrane (1), we suppose that the light-induced conformational change of pHtrII is tightly coupled to that of ppR in each 1:1 ppR/pHtrII unit in the 2:2 complex through the M (rise) and O (decay) state of ppR. The tightly coupled conformational change of the 1:1 ppR/pHtrII unit induces the relative motion between the two units in the 2:2 complex, but the interunit motion occurs delay to intraunit motion of the 1:1 ppR/pHtrII unit.

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